# 300-K Equation of State and High-Pressure Phase Stability of Al<sub>3</sub>BC<sub>3</sub>

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The lattice parameters of Al<sub>3</sub>BC<sub>3</sub> have been measured at room temperature up to 7.5 GPa using X-ray powder diffraction with synchrotron radiation. From the obtained pressure-volume relation for Al<sub>3</sub>BC<sub>3</sub>, the isothermal bulk modulus of  $B_0 = 153 \pm 6$ GPa and its pressure derivative of  $dB_0/dp = 19 \pm 4$  have been calculated by fitting a Birch-Murnaghan equation of state. Al<sub>3</sub>BC<sub>3</sub> undergoes no structural transformations up to 1800 K over the 2.5 to 5.3-GPa pressure range. © 2000 Academic Press

*Key Words:* Al<sub>3</sub>BC<sub>3</sub>; high pressure; high temperature; equation of state; phase stability; hardness.

### **INTRODUCTION**

In 1964 Matkovich *et al.* (1) discovered yellow plateletshaped crystals of a new ternary phase of the Al-B-C system. The authors assigned the composition  $Al_4B_{1-3}C_4$ with variable boron content to this compound. According to later investigation by Inoue *et al.* (2), the composition of the phase was determined to be  $Al_8B_4C_7$ .

Hillebrecht and Meyer solved the crystal structure of this phase using X-ray single crystal diffraction (3) and determined the composition to be  $Al_3BC_3$ . Interesting features of the structure are linear  $[CBC]^{5-}$  anions which are isoelectronic to  $CO_2$ .  $[CBC]^{5-}$  is the 16-electron system with the highest negative charge.  $Al_3BC_3$  shows a superstructure caused by trigonal bipyramidal coordinated Al1 atoms (Fig. 1).

 $Al_3BC_3$  is isotypic with  $Mg_3BN_3$  (4). Since for  $Mg_3BN_3$  the phase transition to an orthorhombic phase was found to occur at 4 GPa and 1500 K (5), the similar phase transition might be expected for  $Al_3BC_3$ .

The present work was aimed at studying room temperature equation of state of  $Al_3BC_3$  to 7.5 GPa and its thermal phase stability to 1800 K and to 5.3 GPa using energydispersive powder diffraction with synchrotron radiation.

#### EXPERIMENTAL

## Sample Preparation

Transparent, yellow leaflet crystals of  $Al_3BC_3$  were prepared from the elements in molten aluminum according to the method described in (6). The samples were heated under argon atmosphere (molar ratio Al/B/C = 40:2:3) in graphite crucibles to 2200 K for 1 h, followed by cooling at a rate of 100 K h<sup>-1</sup> to 1300 K and then at a rate of 300 K h<sup>-1</sup> to room temperature. The excess aluminum was removed by diluted hydrochloric acid. The by-product, boron containing  $Al_4C_3$ , was separated under a microscope. The  $Al_3BC_3$ single crystals were manually crushed in a boron carbide mortar.

#### High-Pressure High-Temperature X-ray Diffraction

The high-pressure experiments were carried out using the multianvil-type X-ray system MAX80 with anvils of cemented carbide. The diffraction measurements were performed in an energy-dispersive mode at the beam line F2.1 of HASYLAB-DESY (Hamburg). The conditions were 4.5 GeV and 140 to 70 mA of positron current.

The experimental setup has been described earlier (7). The primary polychromatic synchrotron beam was collimated to  $60 \ \mu m$  (height) by  $100 \ \mu m$  (width) and was perpendicular to the vertical axis of the sample chamber. The diffracted beam was collected in the vertical plane using an intrinsic Ge solid-state detector and a Canberra multichannel analyzer.

The detector was calibrated against energy using X-ray fluorescence  $K\alpha$  and  $K\beta$  lines from Rb, Mo, Ag, Ba, and Tb. The diffraction angle,  $\theta = 4.691 \pm 0.001^{\circ}$ , was calculated from diffraction pattern of NaCl taken at ambient conditions.

The temperature of the high-pressure cell was controlled by an Eurotherm PID-regulator within  $\pm 4$  K. The sample





FIG. 1. Crystal structure of Al<sub>3</sub>BC<sub>3</sub> (subcell).

temperature was measured by a Pt10%Rh–Pt thermocouple. The correction for the pressure effect on a thermocouple electromotive force was made using the data of Getting and Kennedy (8). In high-temperature experiments the sample was placed in a graphite ampoule as above 1340 K,  $Al_3BC_3$  reacts with h-BN discs, other setup components, and the thermocouple.

Pressures at different temperatures were determined from the  $d_{002}$  value of hexagonal graphite-like boron nitride using its thermoelastic equation of state (9).

#### Hardness Measurements

The hardness was measured on a MXT70 (Matsuzava) apparatus with a Vickers indenter at loads of 25 and 50 g. Indents were measured on a NU-2E microscope (Carl Zeiss Jena) in a reflected light in the phase contrast mode at  $750 \times$  magnification. The hardness was calculated by the equation  $H_{\rm V} = 0.4636 \times P/a^2$ , where P is load on the indenter and 2a is the indent diagonal.



FIG. 2. Pressure dependence of the lattice parameters of Al<sub>3</sub>BC<sub>3</sub>.

#### **RESULTS AND DISCUSSION**

Synchrotron radiation diffraction patterns of the Al<sub>3</sub>BC<sub>3</sub> subcell exhibit broadened reflections: (100), (101), (006), (103), (104), (105), (008), (106), (107), (110), and occasionally (114), (109), (201), (116), (204), (118). Lattice parameters of the subcell at ambient pressure were calculated to be  $a = 3.401 \pm 0.003$  Å and  $c = 15.84 \pm 0.02$  Å, in good agreement with those reported in (3).



FIG. 3. Pressure-volume data for  $Al_3BC_3$  at 300 K. Dashed line is the Birch-Murnaghan isotherm that is obtained from a least-squares fits to our measurements.



FIG. 4. Diffraction patterns of  $Al_3BC_3$  taken during heating at 5.3 GPa.

The pressure dependencies of Al<sub>3</sub>BC<sub>3</sub> lattice parameters at room temperature are plotted in Fig. 2. The ratio between linear coefficients of Al<sub>3</sub>BC<sub>3</sub> compressibility toward the *c* and *a* axes is  $k_c/k_a = 1.06 \pm 0.04$ , which points to the absence of anisotropy of the Al<sub>3</sub>BC<sub>3</sub> lattice in terms of elastic properties.

A nonlinear three-parameter least-squares fit to the experimental p-V data using the Birch-Murnaghan (thirdorder Eulerian finite-strain) equation of state (10) gave zero-pressure value of bulk modulus  $B_0 = 153 \pm 6$  GPa and its pressure derivative  $dB_0/dp = 19 \pm 4$ . Experimental and calculated values of Al<sub>3</sub>BC<sub>3</sub> volume are given in Fig. 3.

A relatively high value of bulk modulus of  $Al_3BC_3$  allowed one to expect a high hardness of this phase. However, Vickers hardness of  $Al_3BC_3$  single crystals has been found to be rather low: 20.7 GPa under a 25-g load and 18.2 GPa under a 50-g load. For comparison, hardness of  $Al_3C_2B_{48}$  and  $\alpha$ -AlB<sub>12</sub> at a 200-g load are 33 and 27 GPa, respectively (11).

With the aim to study the phase stability of  $Al_3BC_3$  at high pressures and temperatures, diffraction patterns from the sample were collected at a fixed pressure (2.5, 3.7 and 5.3 GPa) and temperatures that increased stepwise in 50 K increments. From the diffraction patterns (characteristic sequence taken at 5.3 GPa is presented in Fig. 4), it is seen that the temperature increase causes the ordering of  $Al_3BC_3$ structure, which manifests itself in considerable decrease of the lines' width. No formation of new crystalline phase is, however, observed up to 1800 K. Thus, it was found that in spite of a structural similarity of  $Al_3BC_3$  to  $Mg_3BN_3$ , the former undergoes no structural transformations up to 1800 K over the whole pressure range under study.

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